

Stereocontrolled elaboration of quaternary carbon centers involving the asymmetric Michael-type alkylation of chiral imines: an efficient enantioselective access to (+)-vincamine

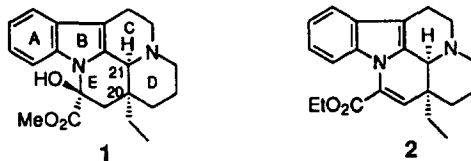
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Abstract: Michael adduct (*S*)-**5b**, resulting from the condensation of chiral imine **4** with methyl acrylate, was transformed in two steps into lactone (*S*)-**8b**. Tryptamine-induced ring-opening of this lactone gave **9**, which was finally converted in three steps into the key tricyclic derivative (*S*)-**11b**, a known precursor of (+)-vincamine **1**. © 1997 Published by Elsevier Science Ltd

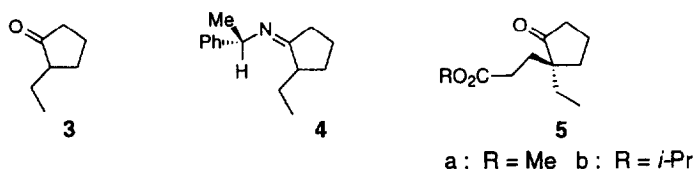
(+)-Vincamine **1** is the major alkaloid encountered in periwinkle (*Vinca minor* L., *Apocynaceae*).¹ In man, the best established pharmacological property of **1**, and semisynthetic derivatives, such as ethyl apovincaminatate **2** (vinpocetine, Cavinton®),² is the cerebroprotective activity, caused by a dilation of cerebral arteries, improving the global cerebral blood flow. Recent investigations have shown that vinpocetine **2** exhibits also a protective effect against brain damage caused by ischemia,³ a gastroprotective action,⁴ and a remarkable promising activity in removing tumoral calcinosis.⁵



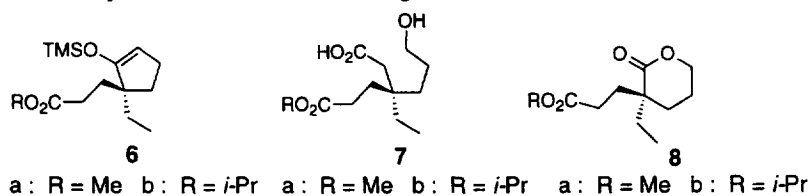
Several strategies for the synthesis of (+)-vincamine **1** (the “biogenetic” numbering system was used here)⁶ and analogs have been developed.¹ The common feature in these approaches was to establish first the tetracyclic [ABCD]-type octahydro[2,3-a]quinolizine system of these alkaloids bearing two controlled stereogenic centers, namely the quaternary carbon center at C-20 and the adjacent methine at C-21 (exemplified by **12**, *vide infra*), and to achieve the synthesis by creating the fifth ring E.

The present basic strategy illustrated a general methodology for the stereocontrolled elaboration of a quaternary carbon center,⁷ as key step in the construction of the tricyclic [ABD]-type cornerstone (*S*)-**11b**, precursor of **12**. Thus, the first stage in the synthesis of our goal **1** was the construction of the disubstituted cyclopentanone **5a**, bearing a stereocontrolled quaternary carbon center (future C-20 center of **1**), by using the asymmetric Michael reaction involving chiral imines we reported a decade ago.⁸ Accordingly, imine **4**, prepared from 2-ethylcyclopentanone **3** and *R*-(+)-1-phenylethylamine of 97% ee (powdered 4 Å molecular sieves, cyclohexane, 20°C, 72 h), was added to methyl acrylate (neat, 24 h at 40°C) to lead, after hydrolytic work-up (10% aqueous AcOH, 20°C, 2 h), to ketoester (*S*)-**5a** (83% overall yield, 90% ee).⁹

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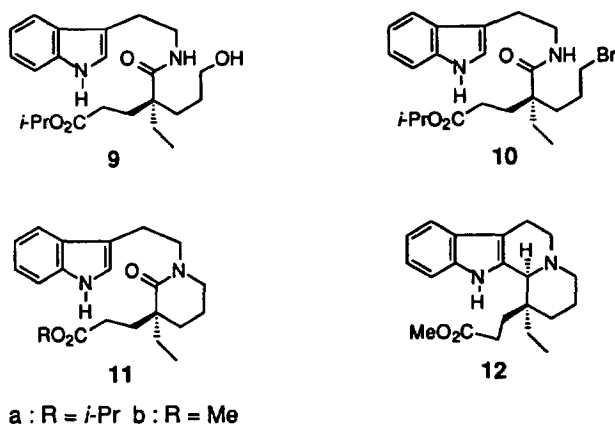


Strategically, we envisaged the construction of the tricyclic key intermediate **11a** by tryptamine-induced ring-opening of the lactone **8a**, derived from adduct **5a**, followed by cyclization into the desired lactam. To attain this end, **5a** was first converted into silyl enol ether (*S*)-**6a**¹⁰ (NaI, TMSCl, Et₃N, 30 min at 20°C in MeCN, 95% yield), which was then transformed into hydroxy acid (*S*)-**7a** (*i*: O₃, CH₂Cl₂/MeOH, -78°C; *ii*: NaBH₄), next cyclized into lactone (*S*)-**8a** (cat. 12 *N* HCl, CH₂Cl₂, 3 h at 20°C, 80% overall yield). However, all attempts at ring-opening of **8a**, by means of tryptamine, were thwarted by the competitive attack to the ester group of the propanoate appendage. We therefore decided to replace the methyl ester group of **8a** by an isopropyl moiety, less sensitive to nucleophilic attack. For that purpose adduct **5a** was transesterified into (*S*)-**5b** (cat. *i*-PrONa, *i*-PrOH, 15 min at 20°C, 80% yield). The latter compound was then transformed into lactone (*S*)-**8b**¹¹ with a 75% yield, according to the same protocol used for converting **5a** into **8a**.



To our delight ring-opening of lactone **8b** now proceeded in a completely chemoselective fashion (DMF, 2 h at 100°C), furnishing compound (*S*)-**9** with a 80% yield.¹¹

We next examined the conversion of amido alcohol **9** into our goal **11b**. With this aim in view, compound **9** was first transformed into bromide (*R*)-**10** (CBr₄/PPh₃, CH₂Cl₂, 2 h at 20°C, 80% yield).¹² Cyclization of **10** was next attempted. After extensive experimentation,¹³ we discovered that treatment of **10** with KH in the presence of 18-crown-6 (THF, 10 min at 20°C) furnished quantitatively the desired tricyclic lactam (*S*)-**11a**. The latter compound, upon transesterification (cat. MeONa, MeOH, 1 h at 20°C, quantitative), finally gave our goal (*S*)-**11b**.¹⁴



Physical and spectroscopic data of compound **11b** proved to be identical in all respects with those reported in the literature.¹⁵ Moreover, seeing that Bischler–Napieralski cyclization of **11b**, prepared by another route,¹⁵ followed by reduction, led to the tetracyclic derivative **12**, a known, direct precursor of (+)-vincamine **1**, the present approach constitutes a formal, efficient enantioselective synthesis of

this alkaloid. Further extensions of the present methodology are currently under investigation in our laboratories.

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References

1. Eburnamonine–Vincamine Alkaloids: Lounasmaa, M.; Tolvalen, A. in *The Alkaloids*; Cordell, G. A., Ed.; Academic Press: New York, **1992**, Vol. 42, p. 1. Saxton, J. E. *Nat. Prod. Rep.* **1996**, *13*, 327–363, and references cited therein.
2. Kawashima, Y.; Ikemoto, T.; Horiguchi, A.; Hayashi, M.; Matsumoto, K.; Kawarasaki, K.; Yamazaki, R.; Okuyama, S.; Hatayama, K. *J. Med. Chem.* **1993**, *36*, 815–819.
3. Krieglstein, J.; Rischke, R. *Eur. J. Pharmacol.* **1991**, *205*, 7–10. Rischke, R.; Krieglstein, J. *Japan J. Pharmacol.* **1991**, *56*, 349–356. Katsura, M.; Kuriyama, K. *Eur. J. Pharmacol.* **1992**, *224*, 117–124.
4. Nosálová, V.; Machová, J.; Babulová, A. *Arzneim.-Forsch. (Drug Res.)* **1993**, *43*, 981–985.
5. Ueyoshi, A.; Ota, K. *J. Int. Med. Res.* **1992**, *20*, 435–443.
6. Le Men, J.; Taylor, W. I. *Experientia*, **1965**, *21*, 508–510.
7. Review: Fuji, K. *Chem. Rev.* **1993**, *93*, 2037–2066.
8. Reviews: d'Angelo, J.; Desmaële, D.; Dumas, F.; Guingant, A. *Tetrahedron: Asymmetry* **1992**, *3*, 459–505. d'Angelo, J.; Cavé, C.; Desmaële, D.; Dumas, F. *Trends in Organic Synthesis*; Pandalai, G., Ed.; Trivandrum: India, **1993**; Vol. 4, pp 555–615.
9. d'Angelo, J.; Revial, G.; Costa, P. R. R.; Castro, R. N.; Antunes, O. A. C. *Tetrahedron: Asymmetry* **1991**, *2*, 199–202. Compound **5a** has been used in a formal synthesis of vallesamidine: Costa, P. R. R.; Castro, R. N.; Farias, F. M. C.; Antunes, O. A. C.; Bergter, L. *Tetrahedron: Asymmetry* **1993**, *4*, 1499–1500.
10. Walshe, N. D. A.; Goodwin, G. B. T.; Smith, G. C.; Woodward, F. E. *Org. Synth.* **1987**, *65*, 1–5.
11. Lower yields were obtained by using Takano's conditions: Takano, S.; Yonaga, M.; Morimoto, M.; Ogasawara, K. *J. Chem. Soc. Perkin Trans. (I)* **1985**, 305–309.
12. Falorni, M.; Lardicci, L.; Giacomelli, G. *J. Org. Chem.* **1986**, *51*, 5291–5294.
13. Lower yields were obtained by using Schlessinger's conditions: Hermann, J. L.; Cregge, R. J.; Richman, J. E.; Kieczkowski, G. R.; Normandin, S. N.; Quesada, M. L.; Semmelhack, C. L.; Poss, A. J.; Schlessinger, R. H. *J. Am. Chem. Soc.* **1979**, *101*, 1540–1544.
14. **5b**: oil; $[\alpha]_{\text{D}}^{20} +9.5$ (c 2, EtOH); IR (cm^{-1}) 2971, 1734; $^1\text{H NMR}$ (200 MHz, CDCl_3) δ 0.84 (t, $J=7.6$ Hz, 3H) 1.23 (d, $J=6.3$ Hz, 6H) 1.39–1.54 (m, 2H) 1.62–2.00 (m, 6H) 2.10–2.40 (m, 4H) 4.99 (m, 1H); $^{13}\text{C NMR}$ (50 MHz, CDCl_3) δ 8.0 (CH_3) 18.2 (CH_2) 21.5 (CH_3) 26.6 (CH_2) 28.6 (CH_2) 29.2 (CH_2) 32.9 (CH_2) 37.9 (CH_2) 50.7 (C) 67.29 (CH) 172.6 (C) 222.0 (C); MS m/z (%) 226 (M^+ , 19) 167 (86) 111 (100); HRMS, calculated for $\text{C}_{13}\text{H}_{22}\text{O}_3$: 226.156895, found: 226.15649. **6b**: oil; IR (cm^{-1}) 1730, 1645; $^1\text{H NMR}$ (200 MHz, C_6D_6) δ 0.12 (s, 9H) 0.84 (t, $J=7.35$ Hz, 3H) 1.03 (d, $J=6.30$ Hz, 3H) 1.04 (d, $J=6.30$ Hz, 3H) 1.27–1.49 (m, 2H) 1.56 (m, 2H) 1.84–1.95 (m, 2H) 2.06 (m, 2H) 2.36 (m, 2H) 4.51 (m, 1H) 5.02 (m, 1H). **8b**: oil; IR (cm^{-1}) 1720; $^1\text{H NMR}$ (200 MHz, CDCl_3) δ 0.92 (t, $J=7.5$ Hz, 3H) 1.22 (d, $J=6.3$ Hz, 6H) 1.45–2.00 (m, 8H) 2.34 (m, 2H) 4.32 (m, 2H) 4.99 (m, 1H); $^{13}\text{C NMR}$ (50 MHz, CDCl_3) δ 8.2 (CH_3) 20.7 (CH_2) 21.4 (2 CH_3) 29.0 (CH_2) 29.5 (CH_2) 31.1 (CH_2) 33.0 (CH_2) 45.1 (C) 67.5 (CH) 69.8 (CH_2) 172.3 (C) 174.8 (C); MS m/z (%) 214 (9) 183 (80) 128 (100). **9**: oil; $[\alpha]_{\text{D}}^{20} -1.1$ (c 4.5, EtOH); IR (cm^{-1}) 3340, 2940, 1720, 1640; $^1\text{H NMR}$ (200 MHz, CDCl_3) δ 0.76 (t, $J=7.40$ Hz, 3H) 1.22 (d, $J=6.22$, 6H) 1.25–1.85 (m, 9H) 2.13 (m, 2H) 2.98 (t, $J=6.60$, 2H) 3.47 (t, $J=6.0$ Hz, 2H) 3.63 (dd, $J=12.5$ Hz, $J=6.0$ Hz, 2H) 4.97 (m, $J=6.2$ Hz, 1H) 5.86 (NH) 7.04–7.28 (m, 3H) 7.38 (d, $J=7.50$ Hz, 1H) 7.63

- (d, $J=7.40$ Hz, 1H) 8.17 (NH); ^{13}C NMR (50 MHz, CDCl_3) δ 8.0 (CH_3) 21.6 (CH_3) 26.8 (CH_2) 29.0 (CH_2) 29.3 (CH_2) 30.0 (CH_2) 39.6 (CH_2) 47.7 (C) 62.3 (CH_2) 67.7 (CH) 111.2 (CH) 112.2 (C) 118.4 (CH) 119.00 (CH) 121.7 (CH) 122.2 (CH) 127.0 (C) 136.3 (C) 173.1 (C) 176.0 (C); MS m/z (%) 402 (M^+ , 4) 254 (15) 143 (100). **10**: oil; $[\alpha]_{\text{D}}^{20} +3.8$ (c 2.1, EtOH); IR (cm^{-1}) 3290, 2971, 1733, 1612.; ^1H NMR (200 MHz, CDCl_3) δ 0.76 (t, $J=7.2$ Hz, 3H) 1.22 (d, $J=6.2$ Hz, 6H) 1.46 (q, $J=7.2$ Hz, 2H) 1.50–1.70 (m, 4H) 1.54–1.82 (m, 2H) 2.04–2.18 (m, 2H) 2.98 (t, $J=6.6$ Hz, 2H) 3.28 (t, $J=6.0$ Hz, 2H) 3.62 (m, 2H) 4.97 (m, 1H) 5.79 (NH) 7.04–7.06 (m, 1H) 7.09–7.25 (m, 2H) 7.38 (m, 1H) 7.62 (m, 1H) 8.23 (NH); ^{13}C NMR (50 MHz, CDCl_3) δ 7.9 (CH_3) 21.6 (2 CH_3) 25.2 (CH_2) 26.7 (CH_2) 27.1 (CH_2) 29.0 (CH_2) 29.3 (CH_2) 32.7 (CH_2) 33.9 (CH_2) 39.5 (CH_2) 47.7 (C) 67.7 (CH) 111.2 (CH) 112.6 (C) 118.5 (CH) 119.3 (CH) 122.0 (2 CH) 127.1 (C) 136.3 (C) 172.8 (C) 175.2 (C). **11a** : oil; $[\alpha]_{\text{D}}^{20} +10.3$ (c 1.55, EtOH); IR (cm^{-1}) 3268 2936 1729 1614; ^1H NMR (200 MHz, CDCl_3) δ 0.86 (t, $J=7.5$, 3H) 1.23 (d, $J=6.2$, 6H) 1.40–2.10 (m, 8H) 2.16–2.42 (m, 2H) 3.00 (m, 2H) 3.20 (m, 2H) 3.65 (m, 2H) 5.00 (m, $J=6.2$ Hz, 1H) 7.01–7.05 (m, 1H) 7.06–7.23 (m, 2H) 7.33–7.40 (m, 1H) 7.61–7.71 (m, 1H) 8.48 (NH); ^{13}C NMR (50 MHz, CDCl_3) δ 8.4 (CH_3) 19.6 (CH_2) 21.6 (CH_3) 22.9 (2 CH_2) 29.3 (CH_2) 30.8 (CH_2) 33.1 (CH_2) 44.1 (CH) 48.4 (CH_2) 48.7 (CH_2) 67.4 (CH) 111.1 (CH) 112.8 (C) 118.5 (CH) 119.0 (CH) 121.6 (CH) 122.1 (CH) 127.3 (C) 136.2 (C) 173.3 (C) 173.6 (C); MS m/z (%) 384 (M^+ , 6) 325 (9) 143 (100).
15. Mekouar, K.; Ambroise, L.; Desmaële, D.; d'Angelo, J. *Synlett*, Special Issue, May **1995**, 529–532. Desmaële, D.; Mekouar, K.; d'Angelo, J., *J. Org. Chem.* **1997**, 62, in press.

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